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Improved *in situ* Sr isotopic analysis by a 257 nm femtosecond laser in combination with the addition of nitrogen for geological minerals

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ABSTRACT

In situ Sr isotope analysis of geological materials by laser ablation multiple collector-inductively coupled plasma mass spectrometry (LA-MC-ICP-MS) is a powerful tracer technique for tracking magmatic source components and geological processes. However, the accuracy and precision of the ⁸⁷Sr/⁸⁶Sr ratio are limited in the analysis of natural minerals because of the low-Sr concentration, the isobaric interference or small grains with complex textural contexts, especially for transparent minerals such as feldspars. In this study, analytical results demonstrated that ablation rates in fs laser ablation were consistent for various samples (0.08-0.11 µm per pulse), but those in ns laser ablation were obviously material properties-dependent, such as the rates of 0.026 µm per pulse and 0.144 µm per pulse for feldspar and pyrite, respectively. In addition, at similar energy fluences, the sensitivities of Sr in feldspars analyzed by the fs laser were 3.4 times higher than those analyzed by the ns laser due to the higher ablation efficiency of the fs pulse. These advantages of the fs laser not only offer the benefit of eliminating or weakening the matrix effect during the laser ablation processes but also help to improve the analytical precision for transparent minerals. We also demonstrated that the isobaric interferences of calcium dimers and argides (CaAr⁺ + CaCa⁺) and Kr⁺ were dramatically reduced by factors of 6.5–11.7 and 5–12.5 in the presence of 6–12 ml min⁻¹ N₂, respectively. Furthermore, with the addition of N₂ (12 ml min⁻¹), the sensitivity of Rb was inhibited, resulting in a decrease of 1.47 times in Rb/Sr signal ratios. Due to the effect of suppressing interferences by adding N₂, both the stability and accuracy of the ⁸⁷Sr/⁸⁶Sr and ⁸⁴Sr/⁸⁶Sr ratios show improvement, especially for the Rb-rich feldspars.

Combining the advantages of the fs laser system with the addition of nitrogen, an improved *in situ* Sr isotope analytical method is then developed. The satisfactory accuracy and precision of the ⁸⁷Sr/⁸⁶Sr ratio from natural plagioclases, a K-feldspar with high Rb/Sr ratios (0.46) and a low-Sr clinopyroxene were obtained, demonstrating the reliability of the proposed method. Four feldspars, which have different contents of the major elements, Sr and Rb, showed homogeneous Sr isotope compositions and were recommended as potential suitable reference materials for *in situ* Sr isotope analysis. As an application, two plagioclases in mafic microgranular enclaves (MMEs) with small grain sizes (200–300 μ m) and wide ranges of Rb/Sr ratios were analyzed and showed obvious variations of the ⁸⁷Sr/⁸⁶Sr ratios from core to rim, which indicated that the proposed method in this study can provide high spatial resolution geochemical information for a single mineral.

1. Introduction

The radiogenic ⁸⁷Sr/⁸⁶Sr ratio is an important geochemical tracer in solid earth sciences (Davidson et al., 2001; Jackson and Hart, 2006). High precision and bulk-rock Sr isotopic ratios have been determined mainly by thermal ionization mass spectrometry (TIMS) (Yang et al., 2010; Koornneef et al., 2015) and multiple collector-inductively coupled plasma mass spectrometry (MC-ICP-MS) (Waight et al., 2002;

Galler et al., 2007; Yang et al., 2011a). However, bulk-rock Sr isotopic ratios generally reflect a mixed signature of various end-members that might not reach isotope equilibrium in the targeted rock (Davidson et al., 2001; Davidson et al., 2007; Ramos and Tepley, 2008). In this regard, *in situ* Sr isotopic measurements using laser ablation (LA)-MC-ICP-MS offer the spatial resolution to identify and distinguish the interand intra-crystalline isotopic variations of mineral grains on the scale of tens of microns. As a result, LA-MC-ICP-MS for Sr isotopic

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measurements has received increasing interest in earth science, environmental science and archaeology applications, including: (1) magmatic minerals such as carbonate, apatite, plagioclase, clinopyroxene, and perovskite (Christensen et al., 1995; Davidson et al., 2001; Waight et al., 2002; Bizzarro et al., 2003; Schmidberger et al., 2003; Ramos et al., 2004; Woodhead et al., 2005; Yang et al., 2009; Kimura et al., 2013; Tong et al., 2016) and melt inclusion (Jackson and Hart, 2006); (2) biogenic carbonate such as otoliths (Outridge et al., 2002; Barnettjohnson et al., 2005) and shells (Christensen et al., 1995; Ramos et al., 2004; Horstwood et al., 2008); and (3) archaeological tooth enamel (Balter et al., 2008; Copeland et al., 2010).

Christensen et al. (1995) first demonstrated the feasibility of in situ Sr isotopic analysis using LA-MC-ICP-MS on both carbonate and feldspar samples. Further studies reported highly precise (30-100 ppm) and accurate ⁸⁷Sr/86Sr isotope ratios with the nanosecond (ns) laser ablation system (Davidson et al., 2001; Waight et al., 2002; Ramos and Tepley, 2008; Kimura et al., 2013; Chen et al., 2015). Most of interesting and measured minerals for in situ Sr isotopic analysis were transparent minerals. The previous studies reported that the interaction between transparent minerals and ns laser pulses was a non-linear absorption process known as avalanche ionization (Liu et al., 1997; Shaheen et al., 2012). In this process, ablating transparent materials by ns laser pulses depends on the pre-existence of seed electrons, which are metallic impurities or thermal ionizations of shallow energy levels in the transparent materials used to initiate laser-induced breakdown. However, the seed electrons are normally only present at low concentrations and are randomly distributed in transparent materials, resulting in stochastic and inefficient ablation behavior (Liu et al., 1997). In the previous studies, therefore, large spot sizes (100–300 μ m) were required to obtain sufficient signal intensity for ⁸⁷Sr/⁸⁶Sr isotopic analyses of geological samples.

Recently, femtosecond (fs) laser ablation systems with short pulse widths have become available (Russo et al., 2002; Hergenröder et al., 2006; Fernández et al., 2007; Pisonero and Günther, 2008; Shaheen et al., 2012). With fs laser ablation, the pulse duration (< 1 ps) is shorter than the phonon relaxation time (ca. 10 ps) (von der Linde et al., 1997; Mao et al., 2004). Ultrafast energy deposition with fs laser pulses vaporizes and ablates the illuminated volume before thermal relaxation sets in. The heat diffusion, which occurs during the ns laser pulse and results in a significant heat affected zone, is significantly inhibited (von der Linde et al., 1997; Mao et al., 2004). In addition, the pulse intensity (irradiance or power density) reaches much larger values for fs pulses $(10^{14}-10^{15} \text{ W cm}^{-2})$ compared to ns pulses (10⁹–10¹⁰ W cm⁻²), which would improve the analytical quality of ablated materials. Due to these advantages, fs-LA-ICP-MS has been successfully applied to the analyses of element ratios and element concentrations (Poitrasson et al., 2003; Bian et al., 2006; Koch et al., 2006; Horn and von Blanckenburg, 2007; Borisova et al., 2008; Jochum et al., 2014; Li et al., 2015; Li et al., 2016) as well as isotope ratios, such as Sr (Campos-Alvarez et al., 2010; Yang et al., 2011b), U-Pb (Freydier et al., 2008; Hirata and Kon, 2008; Kimura et al., 2014), Pb (Shaheen and Fryer, 2010; Chen et al., 2014; Ohata et al., 2015), Mg (Oeser et al., 2014), Si (Chmeleff et al., 2008; Steinhoefel et al., 2011; Schuessler and von Blanckenburg, 2014), Fe (Horn et al., 2006; Steinhoefel et al., 2009b; Steinhoefel et al., 2009a; Oeser et al., 2014), and Cu (Ikehata et al., 2008; Ikehata and Hirata, 2013). However, the detailed studies for ablating transparent minerals by fs laser are not sufficient. There is great potential in elemental quantitative and isotopic ratio analyses with high spatial resolution for transparent minerals using a fs laser system.

An additional challenge for Sr isotope analysis using LA sampling is the variety of isobaric interferences on the Sr isotope spectrum, including Rb, Kr, doubly charged rare earth elements (REEs), calcium dimers and argides and possibly oxides (Ramos et al., 2004; Woodhead et al., 2005; Vroon et al., 2008; Yang et al., 2011b; Kimura et al., 2013; Yang et al., 2014). The strategies of eliminating or reducing these isobaric interferences have been widely explored and still attract significant attention. In recent decades, mixed-gas plasma has been extensively investigated to overcome analytical limitations of the Ar plasma, including the addition of nitrogen (N2), hydrogen (H2), methane, carbon-containing solvents and water (Durrant, 1994; Guillong and Heinrich, 2007; Hu et al., 2008; Fliegel et al., 2011; Hu et al., 2012a; Shaheen et al., 2012; Lin et al., 2014; Liu et al., 2014; Xu et al., 2015; Fu et al., 2016; Tong et al., 2016). N₂ is one of the most widely used molecular gases in ICP-MS and has been utilized mainly for the improvement of sensitivity and stability, the decrease in polyatomic interferences and the attenuation of matrix effects. Recent studies show that adding N₂ to the Ar carrier gas can significantly enhance signal intensity during LA-MC-ICP-MS analysis, such as for Hf. Nd and Pb (Hu et al., 2012a; Shaheen et al., 2012; Xu et al., 2015). The N2-related intensity enhancement was attributed to the higher thermal conductivity of N₂, which results in higher plasma temperature and the increased excitation and ionization efficiency of many elements (Hu et al., 2012a; Shaheen et al., 2012). However, the suppression of signal intensity resulting from N2 addition has been observed in some analyses of B, S and Sr (Lin et al., 2014; Fu et al., 2016; Tong et al., 2016). Thus, the enhancement or suppression of signal intensity caused by the N₂ addition is element-specific. On the other hand, the more conclusive observation is that the addition of N2 can reduce polyatomic interferences in the ICP. For example, Durrant (1994) first reported that the addition of approximately 1% (v/v) N2 to the coolant flow or the addition of approximately 12% N_2 to the cell gas reduced CeO⁺/Ce⁺ and ThO⁺/Th⁺ ratios by a factor of 2-3 in LA-ICP-MS. Hu et al. (2008) noted that the addition of 5–10 ml min⁻¹ N₂ to the central channel gas in LA-ICP-MS increased the sensitivity for most of the 65 investigated elements by a factor of 2–3 while reducing the oxide ratios $(ThO^+/$ Th⁺) (by one order of magnitude) and the hydride ratio (ArH^+/Ar^+) (by a factor of 3). Shaheen et al. (2012) showed increased abundances of doubly charged ions $(U^{++}/U$ and $Ce^{++}/Ce)$ but reduced oxide ratios $(UO^+/U^+$ and ThO^+/Th^+) and mass bias of Pb and Tl upon the addition of 5 ml min⁻¹ N₂. Fu et al. (2016) reported that the addition of N₂ efficiently reduces polyatomic interferences (OO⁺, SH⁺ and OOH⁺) for in situ S isotope analyses in sulfides.

Feldspars have high Sr abundances, ubiquitously occur in igneous rocks and exhibit a large range of crystallization temperatures (Gagnevin et al., 2005; Charlier et al., 2006; Davidson et al., 2007). Therefore, feldspars were selected in this study as the main research mineral for in situ Sr isotope analysis. Comparisons of the ablation rates, elemental sensitivities and analytical precision of ⁸⁷Sr/⁸⁶Sr ratios between a 257 nm fs laser and a 193 nm ArF excimer ns laser were performed. In addition, we systematically investigated the effects and limitations of small amounts of N2 mixed to the central channel gas flow of LA-MC-ICP-MS. The suppression of the sensitivities towards Rb and Kr and yields of polyatomic interferences were found with the addition of N₂. Combining the high ablation efficiency of the fs laser with the capacity of suppressing the interference factors by adding N2, the improved in situ Sr isotopic analysis method using the 257 nm fs laser was established. The feasibility and flexibility of the proposed method were demonstrated by analyzing common plagioclases, K-feldspars with high Rb/Sr ratios and low-Sr clinopyroxenes.

2. Analytical procedure

2.1. Instrumentation

In situ Sr isotope analyses were performed on a Neptune Plus MC-ICP-MS (Thermo Fisher Scientific, Bremen, Germany) coupled with two laser ablation systems at the State Key Laboratory of Geological Processes and Mineral Resources (GPMR), China University of Geosciences (Wuhan), China. The Neptune Plus, a double focusing MC-ICP- MS, was equipped with seven fixed electron multiplier ICs, and nine Faraday cups fitted with $10^{11} \Omega$ resistors. In addition, a large dry

Table 1

Summary of the operating parameters for the 193 nm excimer laser, the 257 nm fs laser, the MC-ICP-MS and the desolvation nebulizer system.

Cup-configuration	L4	L3	L2	Ll	С	H1	H2	H3		
	⁸³ Kr	$^{167}{\rm Er}^{++}$	⁸⁴ Sr	⁸⁵ Rb	86 Sr	173 Yb ⁺⁺	⁸⁷ Sr	⁸⁸ Sr		
RF Power				1250 W						
Cool gas flow				16.0 L mi	iñ ¹					
Auxiliary gas flow		0.10 L miñ ¹								
Argon make-up gas flow			0.80 L min ¹							
Helium carrier gas flow			0.50 L min ¹							
Nitrogen gas flow			0, 6, 12 ml miił							
Interface cones			X skimmer cone + Jet sample cone							
Instrument resolution				~ 400 (lov	w)					
Block number				1						
Cycles of each block				160						
Integration Time (s)				0.524 sec						
Laser ablation system										
Laser type		ArF excimer laser			Yb:YAG femtosecond laser					
Wavelength		193 nm				257 nm				
Pulse length		15 ns				300 fs				
Energy density		2.3-13.3 J		0.50-3.85 J cm ²						
Spot size		60, 90.120) μm			45, 60 μm				
Laser frequency		10 Hz				10-200 Hz	<u>r</u>			
Line scanning rate		-				$5 \mu\mathrm{m s}^{-1}$				
Aridus II desolvation nebuliser system										
Membrane temperature			160°C							
Spray chamber temperature	re		110°C							
Sweep gas flow		2.5 L min ¹ Ar								
Sample uptake rate		50 µL min ¹ PFA nebuliser								
Nebuliser flow		0.95 L min ¹								

interface pump (120 m³ hr⁻¹ pumping speed) and a newly designed X skimmer cone and Jet sample cone were used to increase the instrumental sensitivity. The Faraday collector configuration of the mass system was composed of an array from L4 to H3 to monitor Kr, Rb, Er, Yb and Sr (Table 1). Two different laser systems were used: a GeoLas 2005 system (Lambda Physik, Göttingen, Germany), which consisted of a COMPexPro 102 ArF excimer laser (Coherent Inc., Santa Clara, CA, U.S.A.) and a Microlas optical system; an NWR Femto^{UC} femtosecond system (New Wave Research, Fremont, CA, U.S.A.), which consisted of a 300 fs Yb:KGW femtosecond laser amplifier (PHAROS, Light Conversion Ltd., Vilnius, Lithuania) with a wavelength of 257 nm and ESI NWR Femto laser ablation system components. The laser ablation for both laser systems was conducted under a helium atmosphere in the same ablation cell (Geolas 2005) with a volume of ca. 40 cm³, an inlet nozzle (i.d. < 0.5 mm) and a wide outlet (i.d. 1.5 mm), while argon was mixed into the sample-out line down-stream from the ablation chamber prior to entering the torch. To decrease the Tau effect (essentially a delay in the response of the resistors), a new signalsmoothing device (Hu et al., 2012b) was used downstream from the sample cell, which almost eliminated the short-term variability of the signal. Details of the instrumental operating conditions and measurement parameters are summarized in Table 1.

A LA-ICP-MS combination, consisting of an Agilent 7500a ICP-MS (Agilent Technology, Tokyo, Japan) coupled with the ns- or fs-laser ablation systems, was used for the measurements of major and trace element concentrations in natural minerals and reference glasses. Detailed operating conditions for the laser ablation system and the Q-ICP-MS instrument and data reduction are given in Supplementary Information A.

The CETAC Aridus IITM desolvation nebulizer system (CETAC Technologies, Omaha, USA) was used to investigate the influence of interferences on Sr isotopic determination. A series of NIST SRM 987 solutions doped with interfering elements were measured using MC-ICP-MS. The solutions were self-aspirated at an uptake of 100 ml min⁻¹ through the PFA nebulizer and desolvated by the Aridus II system.

The Surface Profiler (P16⁺, KLA-Tencor, California, USA) was used to measure the depth of the ablated crater. Each crater was profiled across the centerline over a range of $200 \,\mu\text{m}$ using a speed of $10 \,\mu\text{m s}^{-1}$. The P16⁺ Surface Profiler uses a diamond stylus (radius = $2 \,\mu\text{m}$). Tests were performed at the Center of Micro-

Fabrication and Characterization (CMFC) of the Wuhan National Laboratory for Optoelectronics (WNLO), China.

2.2. Samples and reagents

The NIST reference glasses SRM 610 and SRM 612, USGS reference glasses BHVO-2G and GSE-1G and MPI-DING reference glasses ATHO-G, StHs6/80-G and T1-G were analyzed.

Three feldspar megacrysts, YG0440 (albite), YG0383 (albite) and YG4301 (anorthite), were collected from Hebei Province (China) and investigated in this study, A K-feldspar megacryst (Tuyk) from a pegmatite deposit in Henan province (China) with a high Rb/Sr ratio (0.46) was used. Feldspar megacrysts were crushed into 2–5 mm fragments and cleaned with deionized water. Some of the fragments were mounted in epoxy resin discs for elemental and Sr isotope analyses using LA-ICP-MS and LA-MC-ICP-MS, respectively. The remaining fragments were milled to < 200 mesh and were then acid-digested for Sr isotope ratio measurements by TIMS. A natural clinopyroxene megacryst (Cpx, HNB-8) with a low Sr concentration (89.2 μ g g⁻¹) was analyzed. The chemical and Sr isotopic compositions of Cpx HNB-8 have been reported by He et al. (2013) and Tong et al. (2016).

Samples of natural pyrite and garnet from the Rock and Mineral Teaching Section in the China University of Geosciences (Wuhan) were used to investigate the difference in the laser rate between ns and fs laser ablation. Their chemical and isotope compositions were not determined.

A Sr isotope-certified reference material (NIST SRM 987) and a Rb isotope-certified reference material (NIST SRM 984) (NIST, Gaithersburg, USA) were prepared using 2% HNO_3 . Single element standard solutions of Ca, Er and Yb (National Center for Analysis and Testing of Steel Materials, China) were used for the experiments with doped interfering elements. Commercially available nitric acid (GR grade) was further purified twice using a DST-1000 acid purification system (Savillex, Eden Prairie, USA).

2.3. LA-MC-ICP-MS measurement and data reduction

For *in situ* Sr isotope measurements, the combination of the highsensitivity X skimmer cone and JET-sample cone was employed, and the mass spectrometer was operated in the low mass resolution mode. A small amount of N_2 (6–12 ml min⁻¹) was added to the carrier gas flow behind the signal-smoothing device by a simple Y connector. NIST SRM 610 was used to optimize the instrumental parameters, including the He and Ar gas flow rates, the torch position, the RF power setting, and the source lens settings for maximum sensitivity and optimum peak flatness. The routine data acquisition consisted of one block of 160 cycles (0.524 s integration time per cycle), with the first 50 cycles being for background collection (no laser ablation) and the remaining 110 cycles for signal collection. The detailed setup parameters of the laser system and MC-ICP-MS are listed in Table 1, including the Faraday-cup setup in the mass spectrometer.

The data reduction for LA-MC-ICP-MS analysis was conducted using Excel spreadsheets. The interference correction strategy was the same as the one reported by Tong et al. (2016). In summary, the regions of integration for both gas background and sample were selected first. Following background correction, which removes the background Kr⁺ signals, no additional Kr peak stripping was applied. Interferences were corrected in the following sequence: (1) the interferences of ¹⁶⁸Er⁺⁺ on ⁸⁴Sr, ¹⁷⁰Er⁺⁺ and ¹⁷⁰Yb⁺⁺ on ⁸⁵Rb, ¹⁷²Yb⁺⁺ on ⁸⁶Sr, and ¹⁷⁴Yb⁺⁺ on ⁸⁷Sr were corrected based on the measured signal intensities of ¹⁶⁷Er⁺⁺, ¹⁷³Yb⁺⁺, ^{84, 86–88}Sr and ⁸⁵Rb and the natural isotope ratios of Er and Yb (Berglund and Wieser, 2011); (2) the isobaric interference of ⁸⁷Rb on ⁸⁷Sr was corrected by monitoring the ⁸⁵Rb signal intensity and a user-specified ⁸⁷Rb/⁸⁵Rb ratio using an exponential law for mass bias. The user-specified ⁸⁷Rb/⁸⁵Rb ratio was calculated by measuring some reference materials with a known ⁸⁷Sr/⁸⁶Sr ratio. Following the

interference corrections, mass fractionation of Sr isotopes was corrected by assuming 88 Sr/ 86 Sr = 8.375209 (Jochum et al., 2009) and applying the exponential law (Russell et al., 1978).

3. Results and discussion

3.1. Ablation rate

In this study, a range of natural minerals (pyrite, garnet, albite YG0383 and Cpx HNB-8) and NIST SRM 610 glass were used to investigate the ablation rates of ns and fs laser pulses. Fig. 1 shows the relationship between the number of laser pulses and crater depth in various samples obtained by ns and fs laser ablation. When using the ns laser at a repetition rate of 10 Hz, a laser spot of 60 µm and a laser fluence of $\sim 9.5 \, \text{J cm}^{-2}$, the depth of ablation craters differed significantly in different types of materials (Fig. 1a). The crater depth in pyrite was significantly higher compared to silicate minerals and NIST SRM 610 glass. For albite YG0383, the lowest ablation rate of 0.026 μm per pulse was found, which was approximately 3.3 times lower than that of NIST SRM 610 (0.086 µm per pulse) (Fig. 1a). In contrast, for fs laser ablation with a repetition rate of 10 Hz, a laser spot of 60 µm and a laser fluence of $\sim 3.8 \text{ J cm}^{-2}$, the ablation rates of all samples were approximately consistent (0.08–0.11 µm per pulse) (Fig. 1b). Moreover, the lower fluence of the fs laser (\sim 3.8 J cm⁻²) contributed the higher ablation rates for feldspar (albite YG0383), Cpx (HNB-8) and garnet, indicating the better ablation efficiency of the fs pulse relative to the ns laser for natural transparency minerals.

Although the 193 nm ultraviolet (UV) can be absorbed by most natural minerals (Günther et al., 1999), there is a fundamental difference in the ablation mechanism between ns and fs laser ablation. For the ns laser, ablation starts through the absorption of the incident laser energy by the free electrons in the target material (Liu et al., 1997; Hergenröder et al., 2006; Shaheen et al., 2012). The ablation rate of the ns laser would be influenced by the physical and chemical properties of the samples (Fig. 1a). Pyrite can effectively absorb laser energy and produce a high ablation rate during ns laser ablation (0.144 µm per pulse) because it contains many free electrons. However, free electrons normally present at low concentrations and randomly distributed in transparent minerals, such as natural silicate minerals, result in a stochastic and inefficient ablation (Liu et al., 1997; Hergenröder et al., 2006; Shaheen et al., 2012). Feldspars are transparent minerals and often lack metal elements, such as Fe. Therefore, albite YG0383 shows the lowest ablation rate (0.026 µm per pulse). For fs laser ablation, due to the high pulse intensity, the interaction of fs laser pulses with solid materials is mainly dominated by two types of non-linear absorption processes: avalanche ionization and multi-photon ionization. In multiphoton ionization, bound electrons of the transparent material can be directly ionized by simultaneously absorbing multiple photons and forming a mass of free electrons to initiate the avalanche ionization. Then, the energy transfer of the fs pulses into the samples does not depend on already pre-existing free electrons, and the ablation event is more deterministic and initiated by a similar laser-material interaction mechanism for various materials with different physical and chemical properties (Liu et al., 1997; Shaheen et al., 2012). Therefore, the ablation rates of the fs laser are consistent for various samples, including pyrite and feldspar (Fig. 1b). This advantage not only shows the benefit of eliminating or weakening the matrix effect during the laser ablation processes but also helps to improve the analytical sensitivity for transparent minerals, such as feldspars.

3.2. Signal responses in ICP-MS

Fig. 2 shows the Sr responses (cps/ μ g g⁻¹) obtained from LA-ICP-MS for reference glasses (NIST SRM 610, NIST SRM 612, ATHO-G, BHVO-2G) and natural feldspars (albites YG0383 and K-feldspar Tuyk) with changing laser fluences of the ns and fs lasers. For the ns laser, Sr responses were obviously different among the NIST glasses, USGS glasses and natural feldspars (Fig. 2), which were consistent with the above studies about ablation characteristics. The Sr responses of natural feldspars obtained by the ns laser were two times lower than those of the NIST glasses at the same fluence. The lower ablation rate of feldspars was considered as the main reason. In contrast, for fs laser ablation, a similar response trend with increasing laser fluence was observed in all the samples (Fig. 2). This is in line with the expectation of the similar ablation rates for various materials ablated by the fs laser, as mentioned above. The direct and exact comparison between the ns laser and fs laser used in this study cannot be achieved due to the different energy distributions (flat peak vs Gaussian peak) and energy outputs. However, as a rough comparison, the Sr responses in feldspars from fs laser ablation at 3.8 J cm^{-2} were 3.4 times higher than those from the ns laser at 4.1 J cm $^{-2}$. In addition to higher ablation rates of the fs laser pulses, the high sensitivity observed in fs laser ablation could be related to the size of laser-generated particles. The particles produced by fs laser ablation have small sizes distributed over a narrow range (D'Abzac et al., 2012, 2013). This improves particle transport efficiency and ionization inside the ICP and consequently improves the sensitivity and signal reproducibility. However, the direct evidences between the improved signal sensitivity and the particle morphology were not provided in this study. Further investigations are needed to validate this assumption.

The promotion of signal intensity benefits the improvement of the precision for isotope ratio measurements. The within-run precisions



Fig. 1. Depth vs applied number of laser pulses for various samples using the 193 nm ns laser system with a repetition rate of 10 Hz, a laser spot of 60 μ m and an energy density of ~9.5 J cm⁻² (a) and using the 257 nm fs laser system with a repetition rate of 10 Hz, a laser spot of 60 μ m and an energy density of ~3.8 J cm⁻² (b).



Fig. 2. The Sr signal responses $(cps/\mu g g^{-1})$ for reference glasses (NIST SRM 610, NSIT SRM 612, ATHO-G, and BHVO-2G) and nature feldspars (albites YG0383 and K-feldspar Tuyk) with changing laser fluences for the 193 nm ns laser (a) and 257 nm fs laser (b). The other laser parameters were the repetition rate of 10 Hz and the laser spot of 60 μ m in both laser ablation system.

(standard error, SE, k = 2) of the ${}^{87}\text{Sr}/{}^{86}\text{Sr}$ ratio in BHVO-2G and natural albite (YG0440) by the ns and fs lasers are shown in Fig. 3. The same laser repetition rate of 10 Hz and a spot size of 60 µm were used, but the fluences were $\sim 9.5 \text{ J cm}^{-2}$ and 3.8 J cm⁻² for the ns laser and fs laser, respectively. The concentrations of Sr and Rb are 396 μ g g⁻¹ and 9.20 μ g g⁻¹ for BHVO-2G and 390 μ g g⁻¹ and 0.23 μ g g⁻¹ for YG0440, respectively. The theoretical precisions, Poisson counting statistics errors (SD_P, Yang et al., 2011b), were calculated. As seen in Fig. 3, the observed 2SE of the 87 Sr/ 86 Sr ratio correlates well with the SD_p . For the fs laser, the ⁸⁸Sr signal intensity and 2SE of the ⁸⁷Sr/⁸⁶Sr ratio were approximately similar for BHVO-2G and YG0440. However, for the ns laser, the ⁸⁸Sr signal intensity of YG0440 was approximately 0.88 V and \sim 42% lower than that of BVHO-2G (1.52 V), resulting in the obvious deterioration of the 2SE of the ⁸⁷Sr/⁸⁶Sr ratio (Fig. 3). Obviously, the fs laser improves the analytical sensitivity and the precision for transparent minerals (feldspars) due to the better ablation efficiency. This advantage will become more apparent for the low-Sr minerals or high spatial resolution cases.

3.3. Influence of N_2 addition on interference correction

LA-MC-ICP-MS is a solid sample direct injection technique. Many interfering elements will be introduced into ICP-MS due to the absence of a chromatographic purification process. The main interferences of Sr isotope analysis for feldspar are Rb, Kr, calcium dimers and argides, and doubly charged Er and Yb (Ramos et al., 2004). In this section, we



Fig. 3. Correlations between the within-run precisions (standard error, SE, k = 2) of single spot analyses and ⁸⁸Sr signal intensities of the reference glass BHVO-2G and natural albite (YG0440) analyzed by ns and fs laser ablation.

describe our investigation of the effect of inhibiting sensitivities of interfering elements and yields of polyatomic ions by the addition of N₂.

3.3.1. Calcium dimers and argides, doubly charged Er and Yb

In this study, NIST SRM 987 solutions (100 μ g l⁻¹) doped with interfering elements (Er and Yb) and a Ca solution (Ca = $500 \ \mu g l^{-1}$) were measured using MC-ICP-MS with a desolvation sample introduction system (Aridus II) and GE on mode. The ion beams on masses of 84, 83.5 and 86.5 were collected to represent the polyatomic ions of Calcium (CaAr⁺/CaCa⁺), doubly charged Er $(167Er^{++})$ and Yb $(^{173}Yb^{+\,+})$, respectively. The signal intensities of $^{88}Sr,\ ^{167}Er^{+\,+},\ ^{173}Yb^{+\,+}$ and CaAr $^+/CaCa^+$ as a function of N_2 addition are illustrated in Fig. 4. Increasing N₂ from 0 ml min^{-1} to 12 ml min^{-1} leads to a gradual decrease in the signal intensity of ⁸⁸Sr. Sr with a low ionization potential (IP = 5.7 eV) can be completely ionized in a typical ICP; therefore, its sensitivity cannot be improved by increasing the temperature of ICP by adding N₂. 167 Er⁺⁺ and 173 Yb⁺⁺ were suppressed at a similar extent of Sr (Fig. 4b and c, respectively). However, the signal intensity of CaAr⁺/CaCa⁺ shows a significant decrease associated with an increase in the N₂ flow rate, such as the signal suppression of 6.5 times and 11.7 times for N₂ additions of 6 ml min⁻¹ and 12 ml min⁻¹, respectively. This may be due to the higher ICP temperature by adding N₂, which could result in decomposition of the polyatomic ion.

3.3.2. Krypton (Kr)

The noble gas krypton (Kr) interferes with masses ⁸⁴Sr and ⁸⁶Sr and is therefore an important interference that needs to be corrected. The source of the Kr is the argon (and helium) gas used to transport the sample into the plasma. The amount of krypton in the argon (and helium) gas is supplier dependent. Woodhead et al. (2005) observed that the Kr abundances in the Ar supply vary largely between different batches, with a total Kr contribution of approximately 20 mV. Ramos et al. (2004) reported that Kr is present in low levels as an impurity in the argon gas (typically < 1 mV for ⁸³Kr) and in the helium transfer gas used during laser ablation (< 3 mV for ⁸³Kr). Kimura et al. (2013) suggested that both signal suppression and enhancement are possible on Kr baselines due to (1) the suppression by laser aerosol loading and (2) the enhancement from Kr in the ablated samples.

In this section, the natural albite (YG0383) was analyzed by fs-LA-MC-ICP-MS in GE on mode. Fig. 5 presents the average signal intensities of 88 Sr⁺ from YG0383 and 83 Kr⁺ in the background without laser firing as a function of N₂ addition. With the addition of N₂ from 0 ml min⁻¹ to 6 ml min⁻¹ and 12 ml min⁻¹, the signal intensities of 88 Sr⁺ in YG0383 decreased from 3.3 V to ~2.5 V (1.32 times) and ~1.9 V (1.73 times), respectively (Fig. 5a). However, the signal intensities of 83 Kr⁺ decreased significantly from 0.00163 V to 0.00027 V (6.0 times) and 0.00013 V (12.5 times), respectively (Fig. 5b). Kr has a high first



Fig. 4. The integrated average signal intensities of 88 Sr⁺, 167 Er⁺⁺, 173 Yb⁺⁺ and CaAr⁺/CaCa⁺ from a series of NIST SRM 987 solutions as a function of sweep gas flow rate combined in both normal and N₂ (N₂ = 6–12 ml min⁻¹) modes.

ionization energy (14.0 eV) relative to Sr (4.2 eV), which could theoretically be improved the sensitivity due to the increase in ICP temperature by adding N₂. However, the reverse experimental results in Fig. 5 indicated that another important parameter must be considered. The evaporation enthalpies of Kr and Sr are 9.0 kJ mol⁻¹ and 144 kJ mol⁻¹, respectively. With the increase in ICP temperature by adding N₂, Kr could evaporate and ionize earlier than Sr near the torch, resulting in a separation of the optimum-ICP parameters between Kr and Sr. Then, the shorter sampling depth (the *Z*-axis value) and the higher sample gas flow rate should increase the sensitivity of Kr. Fig. 5b shows that the signal intensities of Kr increase gradually with the sample gas flow rate from 0.10 ml min⁻¹ to 0.58 ml min⁻¹, which is consistent with our speculation. To the best of our knowledge, this study shows for the first time that the addition of nitrogen dramatically suppressed the Kr interference, which would significantly benefit for the *in situ* Sr isotopic analysis by LA-MC-ICP-MS. Further investigations are needed to reveal the exact mechanism behind this phenomenon.

3.3.3. Rubidium (Rb)

Rubidium (Rb) is a well-known interference with ⁸⁷Sr for the ⁸⁷Sr/⁸⁶Sr ratio analysis. Feldspar samples often contain a large amount of Rb, which can substitute for K in the way of isomorphism. The isobaric interference of ⁸⁷Rb with ⁸⁷Sr was corrected by the measured ⁸⁶Sr/⁸⁸Sr ratio and a natural ⁸⁷Rb/⁸⁵Rb ratio of 0.38571 (Ramos et al., 2004). In this method, the mass bias of Rb is assumed to be consistent with that of Sr. However, it has been recognized in recent years that different elements can isotopically fractionate differently in MC-ICP-MS, such as Pb and Tl (Zhang et al., 2016). An alternative approach is using a certified reference material to calibrate a user-specified ⁸⁷Rb/⁸⁵Rb ratio (Jackson and Hart, 2006; Jochum et al., 2009; Tong



Fig. 5. Fig. 5 (a) The average signal intensities of 88 Sr⁺ obtained from YG0383 by fs-laser ablation MC-ICP-MS as a function of N₂ addition (0, 6, 12 ml min⁻¹). (b) The average signal intensities of 83 Kr⁺ in the gas background without laser firing as a function of N₂ addition (0, 6, 12 ml min⁻¹). The vertical lines in (b) represent the optimum range of sample gas flow rate for 88 Sr⁺ signal intensity at different N₂ gas flow rates.



Fig. 6. (a) The accuracy of ⁸⁷Sr/⁸⁶Sr for a series of NIST SRM 987 doped with increasing amounts of Rb at normal and N₂ (N₂ = 12 ml min⁻¹) modes. The interference of ⁸⁷Rb on ⁸⁷Sr was corrected by the natural isotopic composition of Rb and f_{Sr} . (b) The same data was corrected using a user-specified f_{Rb} , which was obtained from a NIST SRM 984 solution (100 µg l⁻¹). Gray fields represent the accuracy of ⁸⁷Sr/⁸⁶Sr < 0.0002. The black lines in each panel are simulated lines with different Δf_{Rb} values.

et al., 2016). The essence of the latter approach is the use of two different mass bias factors (f) for Sr and Rb. Through determining more accurate Rb mass bias factors (f_{Rb}), we were able to measure high Rb/Sr ratio materials.

In this section, we first simulated the effect of Δf_{Rb} at different Rb/Sr ratio conditions, as shown in Fig. 6. Δf_{Rb} is the deviation between the used f_{Rb} and the true f_{Rb} . As seen in Fig. 6a, the black lines represent the accuracies of the ⁸⁷Sr/⁸⁶Sr ratios calibrated by different Δf_{Rb} values from -10% to 10%. If Δf_{Rb} values are > 10% or -10%, the accuracies of the ⁸⁷Sr/⁸⁶Sr ratios will be > 0.0002 for the samples with Rb/Sr ratios > -0.01, while the superior accuracies of < 0.0002 can be obtained for the samples with higher Rb/Sr ratios > 0.10 if the Δf_{Rb} values are smaller than 1%. In other words, the higher Rb/Sr ratio samples can be measured accurately when the used f_{Rb} is closer to the true f_{Rb} .

After the mathematical simulation, we have undertaken a series of tests using Rb-doped (NIST SRM 984 Rb solution) NIST SRM 987 solutions (Fig. 6). In Fig. 6a, the mass bias factor of Sr (f_{Sr}) was used to correct the mass bias of Rb. In the normal condition (N₂ = 0 ml min⁻¹), the deviation of f_{Rb} and f_{Sr} (Δf_{Rb}) was up to ~10%, resulting in the deviation of the ⁸⁷Sr/⁸⁶Sr ratios of > 0.0002 for Rb/Sr ratios > 0.05. The measured results were well overlapped with the simulated lines. In Fig. 6b, the same data in Fig. 6a were processed again using a user-specified f_{Rb} , which was obtained from a Rb standard solution (NIST SRM 984, 100 µg l⁻¹). The results show that the accuracy of the ⁸⁷Sr/⁸⁶Sr ratios significantly improved to < 0.0002, even with Rb/Sr ratio = 0.2. Conclusively, the accuracy of f_{Rb} and the Rb/Sr ratio are important parameters for the exact Rb-interference correction.

An interesting finding was that the sensitivities of Rb in the normal condition ($N_2 = 0 \text{ ml min}^{-1}$) were always higher than those of Sr. The measured Rb/Sr signal ratios were thus higher than the reference Rb/Sr ratios in the samples (Fig. 7). With the addition of N_2 (12 ml min⁻¹), the sensitivities of Rb and Sr suffered an obvious suppression. However, Rb was worse, reducing the Rb/Sr signal ratios by approximately 1.47 times (Fig. 7). The first ionization energy of Rb (5.7 eV) is similar to that of Sr (4.2 eV), but the evaporation enthalpy of Rb (72.2 kJ mol⁻¹) is 2 times lower than that of Sr (144 kJ mol⁻¹). Therefore, the difference of evaporation enthalpies could be the main reason for the different degrees of signal suppression of Rb and Sr with the addition of N₂. The advantage of reducing the Rb/Sr signal ratio with the addition of N2 is the more accurate correction of Rb interference for the rich-Rb samples. This is demonstrated in Fig. 6b; with the addition of N₂ (12 ml min⁻¹), the accuracies of the $\frac{87}{Sr}$ / $\frac{86}{Sr}$ ratios were better than 0.0002 even with Rb/Sr ratio = 0.5.



Fig. 7. Effect of N₂ addition on Rb/Sr signal ratios evaluated by solution-MC-ICP-MS. The data was obtained by analyzed a series of NIST SRM 987 doped with increasing amounts of Rb at normal and N₂ (N₂ = 12 ml min⁻¹) modes.

3.3.4. The effect of adding N_2 for Sr isotope analysis

Three feldspars (YG0383, YG0440, Tuyk) were analyzed by fs-LA-MC-ICP-MS to evaluate the effect of interference suppression by adding N2. Fig. 8 presents the accuracies of the 87Sr/86Sr ratios and the $^{84}\text{Sr}/^{86}\text{Sr}$ ratios as a function of N2. With the addition of N2, both stability and accuracy of the measured ⁸⁷Sr/⁸⁶Sr ratios improved, especially for Tuyk, which is Rb-rich K-feldspar and has a high Rb/Sr ratio (0.46) (Fig. 8a). More significant improvements occurred in the 84 Sr/ 86 Sr ratios. In the normal condition (N₂ = 0 ml min⁻¹), residual analytical biases of ⁸⁴Sr/⁸⁶Sr in three feldspars were observed after onpeak background subtractions and mass-fractionation corrections (Fig. 8b). With the addition of N_2 , the accuracies of the 84 Sr/ 86 Sr ratios improved, indicating that either Ca polyatomic interferences or Kr effects (such as Kr baseline suppression or release of trace Kr from the sample) have been inhibited efficiently. Though the addition of N2 reduces the sensitivity of Sr in both solution and LA modes, higher suppression of Kr, Rb and polyatomic interferences improves the signal-tonoise ratio and thus increases the accuracy and precision of Sr isotope analysis.



Fig. 8. Effect of N₂ addition on the ⁸⁷Sr/⁸⁶Sr ratios (a) and ⁸⁴Sr/⁸⁶Sr ratios (b) for three natural feldspars (YG0383, YG0440, Tuyk). The horizontal black line in Fig. 8(b) represents the natural ratio of ⁸⁴Sr/⁸⁶Sr (0.05657).

4. Application to natural samples

According to the investigation of ablation characteristics of the fs laser and the effects of the addition of N_2 , an improved *in situ* Sr isotopes analytical method was established. The feasibility and flexibility of the proposed method were verified by analyzing various natural samples.

4.1. Feldspar crystals

Four natural feldspars were analyzed, and the results are shown in Table 2 and Fig. 9. The first three feldspars are plagioclases with different An values (CaO / $(2 \times Na_2O + 2 \times K_2O + CaO)$ in mol.%) and Sr concentrations. The concentrations of Rb in the three feldspars are very low ($< 0.67 \,\mu g \, g^{-1}$). The measured results of the 87 Sr/ 86 Sr ratios for the three feldspars using fs-LA-MC-ICP-MS with $N_2 = 12 \text{ ml min}^{-1}$ were 0.713724 ± 0.000030 (YG0440: 0.713718 using TIMS), 0.710909 ± 0.000040 (YG0383: 0.710919 using TIMS) and 0.703421 ± 0.000030 (YG4301: 0.703426 using TIMS). The accuracies of the averaged values were from -0.000014 to 0.000009. The measurement reproducibility, defined as two times of the relative standard deviation (RSD, k = 2) of repeated analyses, was better than 0.000042. In addition, the ⁸⁴Sr/⁸⁶Sr ratios were relatively constant among the three feldspars and similar to the natural ratio of 0.05657 (Tong et al., 2016). These data confirmed the availability of the proposed method for the common and low Rb/Sr ratio plagioclase samples.

The fourth feldspar sample in Table 2 is a K-feldspar phenocryst (Tuyk), which has been used as an in-house reference material for *in situ* Pb isotope analysis (Zhang et al., 2015). The concentrations of Rb and Sr in the Tuyk are 285 μ g g⁻¹ and 620 μ g g⁻¹, respectively. The Rb/Sr ratio is approximately 0.46, which is higher than the critical value of 0.2 defined by Davidson et al. (2001) and Jackson and Hart (2006). In addition to using fs-laser ablation and N₂, we selected three reference glasses with high Rb/Sr ratios, StHs6/80-G (Rb/Sr ratio = 0.06), T1-G (Rb/Sr ratio = 0.28) and GSE-1G (Rb/Sr ratio = 0.80), to calculate the user-specified *f_{Rb}* for correcting Rb interference. The average value for ⁸⁷Sr/⁸⁶Sr acquired by fs-LA-MC-ICP-MS was 0.710325 ± 0.000297

(SD, k = 2), which was significantly higher than the TIMS value (0.710206). The analytical accuracy and reproducibility (RSD, k = 2) were 0.000167 and 0.000417, respectively. Obviously, for samples with high Rb/Sr ratios, the interference of Rb seriously inhibited the accuracy and precision of ⁸⁷Sr/⁸⁶Sr ratio analysis. However, our method presents a significant improvement in the accuracy and reproducibility for high Rb/Sr feldspar samples relative to previous studies. For example, Davidson et al. (2001) reported accuracy and reproducibility were 0.0059 and 0.0011, respectively, for plagioclases with a high Rb/Sr ratio (0.52). Jackson and Hart (2006) reported that the basalt glass samples with high Rb/Sr ratios (0.14) had large errors (0.000505).

The above results also demonstrated that the four feldspars, which have various concentrations of the major elements, Sr and Rb, had homogeneous Sr isotope compositions and thus are suitable as reference materials for *in situ* Sr isotope analysis, including feldspar samples with high Rb/Sr ratios (< 0.5).

Our research interest is to decipher the origin of mafic microgranular enclaves (MMEs) in granitoids. In past decades, several models have been proposed to explain the petrogenesis of these MMEs, including residues after partial melting, xenoliths of the country rocks, cumulates formed by early crystallization, or mafic magma enclaves generated during magma mixing (Davidson et al., 2007). Commonly, MMEs contain lots of plagioclases, which are sensitive to the geochemical variations of the magma chamber. Therefore, tracing the ⁸⁷Sr/⁸⁶Sr ratios that vary from core to rim of those plagioclases can provide unique information for understanding the petrogenesis of the MMEs. However, plagioclases in MMEs usually have small grain sizes (200-300 µm) with wide ranges of Rb/Sr ratios. Thus, it is a great challenge for high-precision Sr isotope measurements. To verify our experimental method, we chose MMEs from the Ganze-Daocheng granitic belt in the Yidun arc terrane as a case to study (Wu et al., 2016). Based on careful petrographic observation, two plagioclase grains from thin sections were analyzed by fs-LA-MC-ICP-MS. The analytical results are listed in Fig. 9. As shown in Fig. 9, each plagioclase has a diameter of approximately 200-400 µm. The line scan mode was used with spot size of $45\,\mu m$, repetition rate of $30\,Hz$ and line scanning speed of $5 \,\mu m \, s^{-1}$. Both plagioclases have distinct ${}^{87}Sr/{}^{86}Sr$

Table 2

Sr isotope ratios for four natural feldspars measured by TIMS and fs-LA-MC-ICP-MS.

	TIMS $(n = 3)$ fs-LA-MC-ICP-MS						An%	Sr	Rb	Rb/Sr
	87Sr/86Sr	87Sr/86Sr	2SD	84Sr/86Sr	2SD	n		$\mu g g^{-1}$		
YG0440	0.713718 ± 34	0.713724	0.000030	0.05636	0.00017	23	9	390	0.23	0.0006
YG0383	0.710919 ± 18	0.710909	0.000040	0.05646	0.00020	32	10	1511	0.00	0.0000
YG4301	0.703426 ± 24	0.703421	0.000030	0.05642	0.00024	23	51	1058	0.67	0.0006
Tuyk	0.710206 ± 11	0.710325	0.000297	0.05641	0.00015	13	0	620	285	0.46



Fig. 9. Sr isotope ratios for four natural feldspars measured by TIMS and fs-LA-MC-ICP-MS. The error bars represent ± 2 standard errors (SE). The gray shadows represent the reference range obtained by TIMS. Ave. val., average value; TIMS, values determined by TIMS.

ratios from core to rim (Fig. 10), which could indicate that the Sr isotopes of the initial plagioclase grain does not in equilibrium with surround melt. Therefore, the variation of Sr isotopes in the same plagioclase grain may record the magma mixing process. Further studies and evidences are required to confirm this assumption. Nevertheless, it demonstrated that the fs-LA-ICP-MS analysis method established in this study can trace the variation of Sr isotopic compositions even in one small feldspar grain and provide a new perspective for exploring magma origin and evolution.

4.2. Clinopyroxene with low Sr concentration

Clinopyroxene (Cpx) is one of the major hosts of Sr in the lithospheric mantle along with plagioclase, and its Sr isotopic composition has been widely used to provide information on metasomatic agents affecting the lithospheric mantle (Schmidberger et al., 2003; Tong et al., 2016). The low content of Sr in Cpx is an important constraint on precise isotopic determination of Sr by LA-MC-ICP-MS. In our fs-based system, the use of high repetition rates (up to 1 kHz) combined with higher energy provides higher ion currents accompanied by improved precision and accuracy. A natural Cpx megacryst (HNB-8, $Sr = 89.2 \ \mu g \ g^{-1}$) which has been analyzed for Sr isotope ratios using TIMS and ns-LA-MC-ICP-MS by Tong et al. (2016), was selected in this study. Fig. 11 shows the variation of the ⁸⁷Sr/⁸⁶Sr ratios in HNB-8 as a function of laser frequency and laser fluence. The line scanning mode for a 60 µm spot size was used. The measured ⁸⁷Sr/⁸⁶Sr ratios obtained at low ablation frequencies (20 Hz and 50 Hz) and a low ablation fluence (1.17 J cm^{-2}) were quite scattered due to the low Sr signal intensity. In contrast, at high frequencies (> 100 Hz) and high fluences $(> 2.06 \text{ J cm}^{-2})$, the measured ${}^{87}\text{Sr}/{}^{86}\text{Sr}$ ratios in HNB-8 were consistent and agree with the value of TIMS within uncertainty. Compared to previous research, the reproducibility of the ⁸⁷Sr/⁸⁶Sr ratios of HNB-8 in this study was 0.0001 (2SD) and 3 times lower than that of ns laser ablation (0.0003, 2SD) (Tong et al., 2016). However, using the line scan mode in this study produced a $60 \times 400 \,\mu\text{m}$ laser trough with an area of 24,000 μ m², which was 2.1 times larger than that produced by ns laser ablation (spot size of 120 µm, 11,304 µm²). Although this comparison is not exact, we can understand that the fs laser has a strong flexibility to address different types of samples.)



Fig. 10. 87 Sr/ 86 Sr ratios of two plagioclases from MMEs determined using fs-LA-MC-ICP-MS combined with the addition of N₂ (12 ml min⁻¹). The line scanning mode was used. The detailed laser parameters are listed in each panel. The back linear regions in each panel are laser ablated troughs.



Fig. 11. ⁸⁷Sr/⁸⁶Sr ratios of Cpx crystal HNB-8 determined by fs-LA-MC-ICP-MS as a function of laser frequency (a) and laser fluence (b). The data for the ⁸⁷Sr/⁸⁶Sr ratios from low frequency and low fluence were not collected to calibrate the average value. The error bars represent ± 2 standard errors (2SE). The black dotted lines represent the solution values (Tong et al., 2016). The data of ns laser ablation for HNB-8 published by Tong et al. (2016) was plotted.

5. Conclusion

In this study, we first compared the differences in ablation characteristics, signal responses and precisions of Sr isotope ratio analyses between ns LA and fs LA. The results indicated that the advantages of fs LA were not only the higher ablation rates relative to ns LA but also the elimination of matrix-dependent ablation behavior, producing consistent ablation rates and signal responses in a variety of materials. The natural transparent minerals that are difficult to ablate by ns laser pulses, such as feldspars, gain the maximum benefit from fs LA. As a rough comparison, the Sr responses in feldspars from fs laser ablation at 3.8 J cm^{-2} were 3.4 times higher than those from the ns laser at 4.1 J cm^{-2} , resulting in a significant improvement of the precision of Sr isotope analysis, especially for the low-Sr minerals or high spatial resolution cases.

We demonstrated that sensitivities of Sr, Rb and Kr and yields of polyatomic interferences (CaAr⁺/CaCa⁺) and doubly charged Er (¹⁶⁷Er⁺⁺) and Yb (¹⁷³Yb⁺⁺) were simultaneously suppressed by the addition of N₂. However, the effect of suppression on Sr was obviously lower than that on Rb, Kr and polyatomic interferences (CaAr⁺/CaCa⁺), resulting in the enhancement of the signal-to-noise ratio. This improved the accuracy of the ⁸⁴Sr/⁸⁶Sr ratios, which are easily interfered with by Ca polyatomic interferences and Kr and the by external precision of the ⁸⁷Sr/⁸⁶Sr ratio analysis. In addition, due to the decrease in the Rb/Sr signal ratio, the samples with a Rb/Sr ratio < 0.5 can be measured within the accuracy of ⁸⁷Sr/⁸⁶Sr < 0.0002 using a user-specified *f*_{Rb}.

Combining the high ablation efficiency of the fs laser with the capacity of suppressing the interference factors by adding N_2 , the improved *in situ* Sr isotopic analysis method using a 257 nm fs laser was established and employed to analyze four natural feldspars, including K-feldspars with high Rb/Sr ratios (0.46). The good accuracy and external reproducibility of ⁸⁷Sr/⁸⁶Sr for the four feldspars not only confirmed the availability of the proposed method but also suggested that the four natural feldspars have homogeneous Sr isotope compositions and are suitable candidates for matrix-matched feldspar reference materials.

Despite the somewhat lower accuracy and precision of laser ablation analyses of Sr isotopes compared with the conventional solution measurements, especially for the small samples with low-Sr concentration or high Rb/Sr ratios, LA-MC-ICP-MS can provide useful information on small-scale Sr isotopic variations in many geological applications. In this study, two plagioclases in mafic microgranular enclaves (MMEs) with small grain sizes (200–300 μ m) and wide ranges of Rb/Sr ratios were analyzed and showed obvious variations of the ⁸⁷Sr/⁸⁶Sr ratios from core to rim. In addition, the high frequency of the fs laser was used to analyze the natural Cpx megacryst with a low-Sr concentration (HNB-8, Sr = $89.2 \,\mu g \, g^{-1}$). The analytical reproducibility of the 87 Sr/ 86 Sr ratios in HNB-8 analyzed by the fs laser was 3 times lower than that of the ns laser. Both cases showed the feasibility and flexibility of the fs-LA-MC-ICP-MS analysis method established in this study.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.chemgeo.2017.12.018.

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